

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
13 June 2002 (13.06.2002)

PCT

(10) International Publication Number
WO 02/46483 A2

(51) International Patent Classification⁷: C22B 11/08,
C12Q 1/68, C07H 21/00

(74) Agent: MCDONNELL, John, J.; McDonnell Bochen
Hulbert & Berghoff, 300 South Wacker Drive, Suite 3200,
Chicago, IL 60606 (US).

(21) International Application Number: PCT/US01/45039

(22) International Filing Date:
30 November 2001 (30.11.2001)

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU,
AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU,
CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,
MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI,
SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU,
ZA, ZW.

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
60/251,715 6 December 2000 (06.12.2000) US

(84) Designated States (*regional*): ARIPO patent (GH, GM,
KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW),
Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM),
European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR,
GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent
(BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,
NE, SN, TD, TG).

(71) Applicant (*for all designated States except US*): NORTH-
WESTERN UNIVERSITY [US/US]; 633 Clark Street,
Evanston, IL 60208 (US).

(72) Inventors; and

(75) Inventors/Applicants (*for US only*): MIRKIN, Chad,
A. [US/US]; 111 16th Street, Wilmette, IL 60091 (US).
PARK, So-Jung [KR/US]; 800 Custer Avenue, Apart-
ment 311, Evanston, IL 60602 (US). JIN, RONGCHAO
[CN/US]; 820 Noyes, Apartment 1E, Evanston, Illinois
60201 (US).

Published;

— without international search report and to be republished
upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guid-
ance Notes on Codes and Abbreviations" appearing at the begin-
ning of each regular issue of the PCT Gazette.

WO 02/46483 A2

(54) Title: SILVER STAIN REMOVAL BY CHEMICAL ETCHING AND SONICATION

(57) Abstract: This invention relates to methods for regenerating spent DNA detection chips for further use. Specifically, this invention relates to a method for removal of silver from used DNA detection chips that employ gold nanoparticle-oligonucleotide conjugate probes and that use silver staining for signal amplification. A first method comprises contacting the used silver stained DNA detection chip with a cyanide etching solution. A second method comprises subjecting the silver stained DNA detection chip having bound gold particles to ultrasound waves for a time sufficient to remove the silver stain. The sonication treatment removed the gold nanoparticles as well.

THIS PAGE BLANK (USPTO)

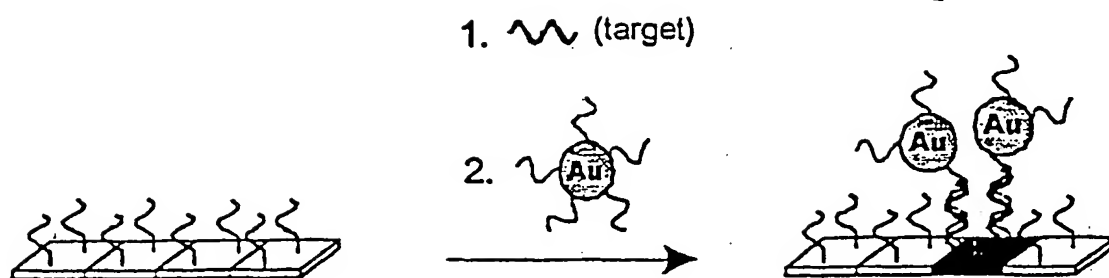


Figure 1

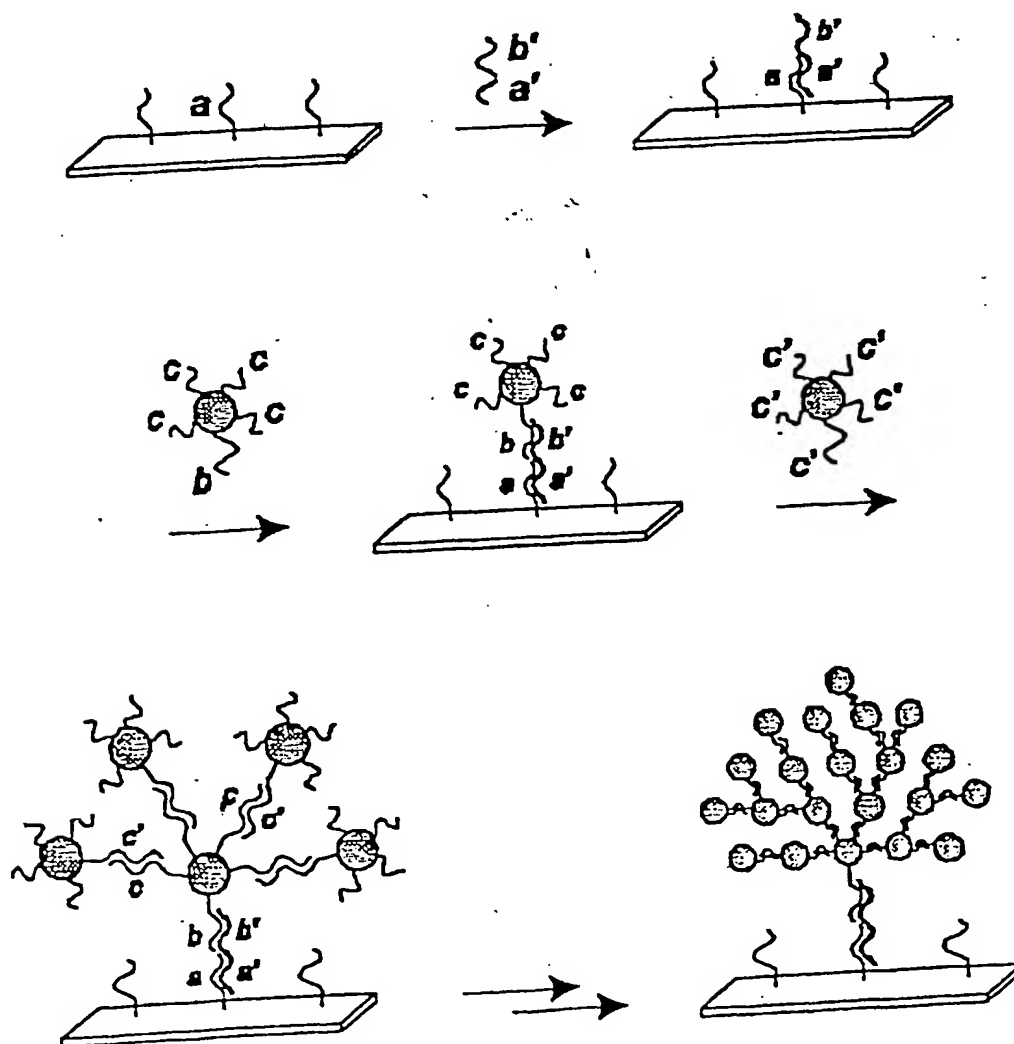


Figure 2

THIS PAGE BLANK (USPTO)

SILVER STAIN REMOVAL BY CHEMICAL ETCHING AND SONICATION

This application claims the benefit of U.S. provisional application number
5 60/251,715, filed December 6, 2000.

FIELD OF THE INVENTION

The present invention relates to method for regenerating spent DNA detection
chips for further use. More specifically, the present invention relates to a method for the
removal of silver from used DNA detection chips that employ gold nanoparticle-
10 oligonucleotide conjugate probes and that use silver staining for signal amplification.

BACKGROUND OF THE INVENTION

Sequence-selective DNA detection has become increasingly important as
scientists unravel the genetic basis of disease and use this new information to improve
medical diagnosis and treatment. DNA hybridization tests on oligonucleotide-modified
15 substrates are commonly used to detect the presence of specific DNA sequences in
solution. The developing promise of combinatorial DNA arrays for probing genetic
information illustrates the importance of these heterogeneous sequence assays to future
science. In most assays, the hybridization of fluorophore-labeled targets to surface bound
probes is monitored by fluorescence microscopy or densitometry. Although fluorescence
20 detection is very sensitive, its use is limited by the expense of the experimental equipment
and by background emissions from most common substrates. In addition, the selectivity
of labeled oligonucleotide targets for perfectly complementary probes over those with
single base mismatches is poor, preventing the use of surface hybridization tests for
detection of single nucleotide polymorphisms. A detection scheme which improved upon
25 the simplicity, sensitivity and selectivity of fluorescent methods could allow the full
potential of combinatorial sequence analysis to be realized.

One such technique is the chip based DNA detection method that employs gold
nanoparticle probes, modified with oligonucleotides, to indicate the presence of a
particular DNA sequence hybridized on a transparent substrate in a three component
30 sandwich assay format. See T.A. Taton, C.A. Mirkin, R.L. Letsinger, *Science*, 289, 1757
(2000). For low target DNA concentrations, the method employs nanoparticle-promoted
silver reduction for signal amplification. The amplification step increases the sensitivity
of the chip and provides for facile DNA detection. However, the silver cannot be readily

removed from the chip by simply washing with water to reuse the chip as one can do with conventional fluorescence-based arrays. Accordingly, a method and composition for regenerating silver stained DNA nanoparticle-based chips for reuse is desirable.

5 BRIEF DESCRIPTION OF THE INVENTION

The present invention relates to compositions and methods for the removal of the silver from used DNA detection chips that employ silver staining for signal amplification. The invention relates to chemical compositions, methods using the chemical compositions and sonication methods for removing the silver.

10 Typically, a plurality of nanoparticle-oligonucleotide conjugates or oligonucleotides can be attached to the substrate in an array for detecting multiple portions of a target nucleic acid, for detecting multiple different nucleic acids, or both. For instance, a substrate may be provided with rows of spots, each spot containing a different type of oligonucleotide or oligonucleotide-nanoparticle conjugate designed to bind to a
15 portion of a target nucleic acid. A sample containing one or more nucleic acids is applied to each spot, and the rest of the assay is performed in one of the ways described above using appropriate oligonucleotide-nanoparticle conjugates, oligonucleotide-liposome conjugates, aggregate probes, core probes, and binding oligonucleotides such as the ones described in WO 98/04740, published February 5, 1998 and WO 00/33079, published
20 June 8, 2000.

When a substrate is employed, a detectable change can be produced or further enhanced by silver staining. Silver staining can be employed with any type of nanoparticles that catalyze the reduction of silver. Preferred are nanoparticles made of noble metals (e.g., gold and silver). See Bassell, et al., *J. Cell. Biol.*, **126**, 863-876
25 (1994); Braun-Howland et al., *Biotechniques*, **13**, 928-931 (1992). If the nanoparticles being employed for the detection of a nucleic acid do not catalyze the reduction of silver, then silver ions can be complexed to the nucleic acid to catalyze the reduction. See Braun et al., *Nature*, **391**, 775 (1998). Also, silver stains are known which can react with the phosphate groups on nucleic acids.

30 Silver staining can be used to produce or enhance a detectable change in any assay performed on a substrate. In particular, silver staining has been found to provide a huge increase in sensitivity for assays employing a single type of nanoparticle, such as the one

illustrated in Figure 1, so that the use of layers of nanoparticles, aggregate probes and core probes can often be eliminated.

In assays for detecting nucleic acids performed on a substrate, the detectable change can be observed with an optical scanner. Suitable scanners include those used to scan documents into a computer which are capable of operating in the reflective mode (e.g., a flatbed scanner), other devices capable of performing this function or which utilize the same types of optics, any type of grayscale-sensitive measurement device, and standard scanners which have been modified to scan substrates according to the invention (e.g., a flatbed scanner modified to include a holder for the substrate). The resolution of the scanner must be sufficient so that the reaction area on the substrate is larger than a single pixel of the scanner. The scanner can be used with any substrate, provided that the detectable change produced by the assay can be observed against the substrate (e.g., a gray spot, such as that produced by silver staining, can be observed against a white background, but cannot be observed against a gray background). The scanner can be a black and white scanner or a color scanner.

A problem associated with the silver enhancement technique is that the silver cannot be simply removed from the combinatorial DNA array substrates or chips that use oligonucleotide-modified gold nanoparticle probes and that employ nanoparticle-promoted silver reduction for signal amplification.

Accordingly, one object of the invention is to provide compositions for the removal of the silver from DNA detection chip, thereby allowing the chip to be recycled and reused.

Another object of the invention is to provide a cyanide-based method for removing silver from the used DNA detection chip allowing for the recycle and reuse of the chip.

Still another object of the invention is to provide a sonication method for the removal of silver and gold nanoparticles from a used DNA detection chip, thus allowing the chip to be recycled and reused.

DETAILED DESCRIPTION OF THE INVENTION

Oligonucleotide-modified gold nanoparticles and unmodified DNA target can be hybridized to oligonucleotide probes attached to a glass substrate in a three component sandwich assay (see Figures 1 and 2). The nanoparticles can either be individual ones

(see Figure 1) or "trees" of nanoparticles (see figure 2). The "trees" increase signal sensitivity as compared to the individual nanoparticles, and the hybridized gold nanoparticles "trees" often can be observed with the naked eye as dark areas on the glass substrate. When "trees" are not used, or to amplify the signal produced by the trees," the hybridized gold nanoparticles can be treated with a silver staining solution. The "trees" accelerate the staining process, making detection of target nucleic acid faster as compared to individual nanoparticles.

A limit to the silver enhancement technique is that the detection chips that employ nanoparticle-promoted silver reduction cannot be recycled because of the difficulty of removing the silver from the chip surface. Accordingly, the present invention provides compositions and methods for the removal of silver from a used DNA detection chip thus allowing for the recycling and reuse of the chip.

The silver is removed from a used silver stained DNA detection chip by dipping the chip in an etching solution. Any etching solution that can remove silver without destroying the oligonucleotide attachments from the support will work. Examples of aqueous solutions that can be used include a KCN solution containing between about 0.1 moles and about 2 moles of KOH per liter of water, preferably about 1 mole per liter of water; and between about 0.05 moles and about 0.5 moles of KCN per liter of water, preferably about 0.1 moles per liter of water. Another example is an aqueous solution that contains between about 0.01 moles and about 0.5 moles of $\text{Na}_2\text{S}_2\text{O}_3$ per liter of water, preferably about 0.1 moles per liter of water; between about 0.1 moles and about 2 moles of KOH per liter of water, preferably about 1 mole per liter of water; between about 0.001 moles and about 0.1 moles of $\text{K}_3\text{Fe}(\text{CN})_6$ per liter of water, preferably about 0.01 moles per liter of water; and between about 0.0001 moles and about 0.005 moles of $\text{K}_4\text{Fe}(\text{CN})_6$ per liter of water, preferably about 0.001 moles per liter of water. Conventional etching solutions such as the ones described in Xia, *et al. Chem. Mater.* 1995, 7, 2332-2337, hereby incorporated by reference, are also useful in practicing the invention.

Preferred etching solutions include: 1 M KOH/0.1 M KCN aqueous solution and 0.1 M $\text{Na}_2\text{S}_2\text{O}_3$ /1.0 M KOH/0.01 M $\text{K}_3\text{Fe}(\text{CN})_6$ /0.001 M $\text{K}_4\text{Fe}(\text{CN})_6$ aqueous solution.

The invention is further illustrated by the following non-limiting examples.

EXAMPLE 1: Assays Employing Silver Staining

Capture oligonucleotides (3'-HS(CH₂)₃-A₁₀ATGCTCAACTCT) were prepared and immobilized on a glass substrate as described in WO 98/04740. A target oligonucleotide (5'-TACGAGTTGAGAATCCTGAATGCG-3', concentrations given below in Table 1 for each experiment) was hybridized with the capture of oligonucleotides in 0.3 M NaCl, 10 mM phosphate buffer as described in WO 98/04740. The substrate was rinsed twice with the same buffer solution and immersed in a solution containing gold nanoparticle probes functionalized with target-complementary DNA (5'-HS(CH₂)₆A₁₀CGCATTTCAGGAT) (preparation described in WO 98/04740) for 12 hours. Next, the substrate was rinsed copiously with 0.3 M NaNO₃ to remove Cl⁻. The substrate was then developed with silver staining solution (1:1 mixture of Silver Enhancer Solutions A and B, Sigma Chemical Co., # S-5020 and # S-5145) for 3 minutes. Grayscale measurements were made by scanning the silver enhanced substrate on a flatbed scanner (normally used for scanning documents into a computer) linked to a computer loaded with software capable of calculating grayscale measurements (e.g., Adobe Photoshop). The results are presented in Table 1 below.

TABLE 1

Target DNA Concentration	Mean Grayscale	Standard Deviation
10 nM	47.27	2.10
5 nM	53.45	0.94
2 nM	54.56	1.17
1 nM	59.98	1.82
500 pM	61.61	2.26
200 pM	90.06	3.71
100 pM	99.04	2.84
50 pM	135.20	7.49
20 pM	155.39	3.66
None (control)	168.16	10.03

EXAMPLE 2: Removal of Silver By Chemical Etching

A silver stained DNA chip, such as the one produced in Example 1, was dipped in ferri/ferrocyanide etchant (0.1 M Na₂S₂O₃, 1.0 M KOH, 0.01 M K₃Fe(CN)₆, 0.001 M K₄Fe(CN)₆). The silver stain and gold nanoparticles were dissolved (by oxidation) and washed away, leaving a transparent DNA chip. The etching time ranges from a few seconds to a few minutes depending on the amount of silver on the chip. To test the regenerated chip, target DNA, gold probes and silver staining solution were applied to the regenerated chip, successively. The regenerated chip worked as well as a new one.

Additionally, the gold nanoparticles hybridized to the regenerated chip could be removed by washing with water or heating above the melting temperature in 0.3 M PBS. These results demonstrate that the chemical etching can remove silver stain without damaging the DNA chip.

5

EXAMPLE 3: Removal of Silver by Sonication

Silver stain and gold nanoparticles could also be removed by sonication. A silver stained DNA chip, such as the one produced in Example 1, was submersed into a sonicator (Branson model No. 2210 sonicator) for 3-5 minutes at room temperature. The sonication process removed the silver metal as well as the gold nanoparticles to produce the regenerated chip. To test the regenerated chip, target DNA, gold probes and silver staining solution were applied to the regenerated chip, successively. The gold nanoparticle signal intensity still maintains above 90 % after 3 cycles. These results demonstrate that the sonication process can remove silver stain without substantial damage to the DNA chip.

15

WHAT WE CLAIM:

1. A method for removing silver from a silver stained DNA detection chip having bound gold nanoparticles comprising the steps of:
 - (a) providing an etching solution;
 - (b) contacting the chip with the etching solution for a time sufficient to remove the silver stain; and
 - (c) washing the etching solution away from the chip.
2. The method according to claim 1 wherein the etching solution is a cyanide etching solution.
3. The method according to claim 2 wherein the cyanide etching solution comprises:
 - 0.01 M to 0.5 M $\text{Na}_2\text{S}_2\text{O}_3$;
 - 0.1 M to 2 M KOH;
 - 0.001 M to 0.1 M $\text{K}_3\text{Fe}(\text{CN})_6$; and
 - 0.0001 M to 0.005 M $\text{K}_4\text{Fe}(\text{CN})_6$.
4. The method of claim 2 wherein the cyanide etching solution comprises:
 - 0.1 M $\text{Na}_2\text{S}_2\text{O}_3$;
 - 1.0 M KOH;
 - 0.01 M $\text{K}_3\text{Fe}(\text{CN})_6$; and
 - 0.001 M $\text{K}_4\text{Fe}(\text{CN})_6$.
5. The method of claim 2 wherein the cyanide etching solution comprises:
 - 0.05 M to 0.5 M KCN; and
 - 0.1 M to 2 M KOH.
6. The method of claim 2 wherein the cyanide etching solution comprises:
 - 0.1 M KCN; and
 - 1 M KOH.

7. The method of claims 3 or 5 wherein the cyanide etching solution is applied by dipping the chip in the cyanide etching solution.
8. The method of claims 3 or 5 wherein the cyanide etching solution is applied by spraying the chip with the cyanide etching solution.
9. The method of claims 3 or 5 wherein the cyanide etching solution is removed by washing with water.
10. The method of claims 3 or 5 wherein the cyanide etching solution is applied for between about 1 second and about 10 minutes.
11. A method for removing silver from a silver stained DNA detection chip having bound gold particles comprising subjecting the chip to ultrasound waves for a time sufficient to remove the silver stain.
12. The method of claim 11, wherein the chip is subjected to ultrasound waves by submersing the chip in a sonicator.
13. The method of claim 12 wherein the chip is submersed in the sonicator for between about 3 minute and about 5 minutes.